

S.S.S. 36 C

DEC 12 - ~~Sub~~ Filtrate from main batch centrifgd. clear, concd. in vac. to dryness.

Wt. dish + subst. 23.4588
 " ~~sub~~ ^{diff} 23.4005
 .0583

23.4626
 05
 ~ after 1 day @ 110

DEC 22 - Diss. in H₂O + x NaOH, neutr., made up to 11.66 cc., centrifgd. from trace insol. matter.

OPTICAL ROTATION

l=1. o=0,0. R_{sp}: 1.14 1.16 1.17 ✓
 $[\alpha]_D = +234^\circ$

$[\alpha]_D = +234^\circ$

Microkjeldahls on 1 cc.

$\frac{1}{50}$ HCl (x1.001)	5.40	5.40	5.00	5.00	Mean: 1.02
" NaOH (x1.015)	4.28	<u>4.34</u>	3.96	<u>4.02</u>	Bl. <u>.22</u>
		1.06		.98	.80

Blanka.

$\frac{1}{50}$ HCl (x1.001)	5.20	5.20	Mean: 5.20
" NaOH (x1.015)	4.92	4.89	<u>4.98</u>
			.22

$N = .8 \times .28 = .224 \text{ mg. N in } 5 \text{ mg.}$

N = 4.5%

HYDROLYSIS

DEC 29 - 2 cc. dil. to 4, 4 cc. 2N HCl added, b. 7 hrs. overnight in ice-bath

DEC 30 - Neutr. w. NaHCO₃. 2 cc. samples used for anal.

	I	II	Bl.
cc. Na ₂ S ₂ O ₃	16.48	16.30	20.04
			<u>16.39</u>
	Mean: 16.39		3.65

$3.65 \times .981 \times .318 = 1.14 \text{ mg. Cu} = 0.6 \text{ mg. gluc.}$

JAN 2 1925 - Hydrol. soln., acidif. w. HCl, gives brownish ppt w. HPW

Redg. Sugars:
24%

Specificity of Original Soln: 1:3,000,000

	Tag Nos.	Tag date	No. flasks	Pring	St
7/9/24	1/188/38	\bar{VI} - 27	11	7/10	7/11
7/10	1/188/38	\bar{VI} - 27, 30	4, 6	7/11	7/12
7/14	1/188, 193/384	\bar{VI} - 3, \bar{VII} - 8	2, 8	7/15	7/16
7/15	1/193/4	\bar{VII} - 8	11	7/16	7/17
7/22	1/193/4	\bar{VII} - 11	9	7/23	7/24
7/23	1/193/4	\bar{VII} - 11	8	7/24	7/25
7/24	1/193/4	\bar{VII} - 11	8	7/25	7/26
9/9	1/193/5	\bar{VIII} - 26	10	9/10	9/11
9/10	1/193/5	\bar{VIII} - 26	9	9/11	9/12
9/19	1/193/6	\bar{IX} - 12	10	9/20	9/22
9/23	1/193/6	\bar{IX} - 12	10, 106	9/24	9/26
9/30	1/193/7	\bar{IX} - 23	10	10/1	10/2
10/1	1/193/7	\bar{IX} - 23	10	10/3	10/4
10/14/24	1/193/8	\bar{X} - 6	8	10/15	10/16
10/16/24	1/193/8	\bar{X} - 6	14	10/17	10/18
10/17/24	1/193/8	\bar{X} - 6	14, 112	10/18	10/20
10/20/24	1/193/8, 9	\bar{X} - 6, 7	1, 13	10/21	10/22
10/21/24	1/193/9	\bar{X} - 7	14	10/22	10/23
10/22	1/193/9, 11	\bar{X} - 7, 15	10, 2	10/23	10/24
10/23	1/193/12, 12	\bar{X} - 15, 22	12, 2	10/24	10/25

then HOAc and 1:1 HCl added until max. ppt. formed. 216 flasks
 Centrgd., ppt. washed 1" w. H₂O.

OCT 27 1924 - Treated as 35A, vol. = 5500 l.

Pptd. w. ca 7 l. alc. sup. mat. - but 1 l. more
 2 alc. added to make sure. OCT 28 - Supernatant
 siphoned off. Ppt. centrgd., taken up again in
 H₂O, centrgd., ppt. washed w. H₂O. 300 ml. at

3 800 cc. vol., pptd. w. 400 cc. alc. & let stand
 overnight in ice-box. OCT 30 - Supernatant, -, siphoned off, ppt.

324
 liters

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centfgd. in cold. ^{aff.} OCT 31 - Radiss. in ca. 500 cc. H₂O, insol. portion
 4 repeatedly centfgd. Main soln. at vol. of 600 cc., pptd. w. 400 cc. alc.
 Ppt., which remained fluid, ~~and~~ taken up in little more H₂O & sald. w. (NH₄)₂SO₄
 but ppt. in this case contained active material as well, so extd. several times w.
 13 H₂O. Solns. dialyzed over night in running H₂O, concd., & pptd. w. alc. ^(2" of rad)
 1 ~~active material to main ppt.~~ Main ppt. taken up in cold H₂O, stirred
 251 thoroughly, dissolved only slightly, so soln., weakly +, added to
 P10 (38). Ppt. again homogenized with 2" frack. above & dil'd to ca.
 250 cc. Set stand on ice over night, centfgd. in cold until
 light ppt. fairly solid. NOV 7 ^{ff.} - Decanted as well as possible, ppt. taken
 5 up again in H₂O, repeatedly centfgd., solns. combined. NOV 12 - Clear
 HOAc but opalescent soln. pptd. w. HOAc to max. ppt. Set stand in ice-box.
 NOV 13 ^{ff.} Centfgd. in cold. Supernatant -, discarded. Ppt. (2 portions)
 washed, ea. portion w. 50 cc. ^{cold} 1/2 HOAc. Centfgd. in cold. Supernatant
 like previous one, gives strong purplish red w. I. Ppt. washed again w. 50 cc.
 ea. 1/2 HOAc. Supernatant -, but still red w. I. NOV 17 - Ea. washed
 again w. 100 cc. 1/2 HOAc, centfgd. in cold. NOV 18 - Supernatant -, now
 gives only very faint color test for glycogen. Poured off, ea. washed
 again w. 50 cc. 1/2 HOAc. ~~gives~~ faint glycogen test after centfy.
 NOV 19 - Ppts. transf. to 1 ~~plate~~ ^{beaker}, dis. by gradual addn. 10% NaOH
 until faintly alk. Repeatedly centfgd. in cold, at vol. of ca.
 6 200 cc. NOV 20 - Clear soln., to which em. amt. remaining of (35)
 HOAc was added, pptd. by shaking w. HOAc in xs. Centfgd.
 NOV 22 ^{ff.} - Supernatant, still somewhat active, at vol. of 125 cc.,
 7 pptd. w. 50 cc. alc. & let stand in cold. Ppts. combined, dis. w.
 NaOH alc. sl. xs. NaOH, & at vol. of 125 cc., pptd. w. xs. alc. Supernatant, still
 somewhat active, ~~neutral~~, concd. to sm. bulk, again made alk., pptd. w.
 xs. alc. added, but no ppt. Acetone then added, pptg salts & 2
 liquid layers. S in salts. These dis. in H₂O & soln. added to that
 of the main ppt. in H₂O. DEC 1 - At vol. of 150 cc., 5 g. NaOAc, 5 cc.
 8 10% NaOH added, then pptd. w. 200 cc. alc. DEC 2 - Mxt. centfgd.
 NaOH, alc. DEC 3 - Supernatant -. Ppt. taken up in H₂O. Soln. still
 yellow. 5 g. NaOAc added. At vol. of 250 cc., after acidification w HOAc,
 9 pptd. w. 75 cc. alc. S acid came down, very voluminous, gelatinous
 form. DEC 5 - Centfgd. Ppt. ansfld. in H₂O, diss. w. sl. xs. NaOH.
 Neutr. again w. HCl, 10 g. NaOAc added, at vol. of 250 cc. acidif. w.
 10 HOAc. Heavy ppt. Set stand, centfgd. Supernatant, -, poured
 HOAc off, ppt. taken up in H₂O. DEC 9 - Diss. by making acid to Congo w.

11
HCl 1:1 HCl. At vol. of 200 cc. pptd. in cold w. 500 cc. cold redistd. alc. Centfgd. in cold. DEC 10 + ~~the~~ Supernatant \pm , discarded. Ppt. taken up in H_2O + few drops $NHCl$. Still somewhat yellowish, so 25 cc. 1:1 HCl added, ~~again~~ + at vol. of 200 cc., again pptd. w. 500 cc. alc. ~~DEC 10~~ Centfgd. Supernatant -, discarded. DEC 11 - Ppt. diss. in ca. 200 cc. H_2O , ^{+ few drops $NHCl$} incl. rinsings, rinsed into 2 collodion bags, diald. agst. running H_2O , then agst. successive changes distd. H_2O + Benzene in ice-box until no Cl^- inside or out. DEC 17 - Transf. to centfg. bottles & centfgd. repeatedly. Supernatant: (37A). Ppt. JAN 5 1925 ^{redistd} Taken up w. alc., centfgd., then washed 2X w. redistd. acetone. JAN 8 - Filtered off, washed w. acetone, dried in warm vacuum desiccator. Yield: 3.2 g.

Active at 1:6,000,000 diln. - at 1,800,000.

at 1:400 gives sl. turbidity w. III serum.

Portion b. few min. w. naphthoresorcin + conc. HCl, cooled, dild, extd. w. CH_2O , gives violet color absorbing in yellow - evidence of glucuronic acid. Also gives this test after treatment w. HNO_2 .

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Wt. bot + air-dry subst 6.8209 Dried in vac at T. P. CHCl_3
 " " 6.6126

6.8209
 6.6995
 .0214

Dis. w. few drops $\text{N HNO}_3 + \text{H}_2\text{O}$,
 conty. w. N NaOH until pptd. & redissolved, made up
 to 20 cc. 6.8015 6.8002 6.7998
 6.6975
 .1023

MICROKJELDAHL'S

2 cc. samples

cc. $\frac{\text{N}}{50} \text{HCl}$ (1.001)	5.00	5.00		
cc. $\frac{\text{N}}{100} \text{NaOH}$ (1.014)	14.00	15.40		
	8.00	9.00	10.01	
	6.00	6.08	6.40	6.48
		3.93		3.53
		.44	4.8% N	.44
		3.49		3.09

4.5% N N
4.5%

OPTICAL ROTATION

BR. -0.04 -0.04 -0.05 Rdp. +1.46 1.46 1.45 $l=1$
 $\frac{1.50 \times 200}{.023}$ $[\alpha]_D = +800.293.3^\circ$

$[\alpha]_D = +295.3^\circ$
Ash-free

ASH

Wt. bot + subst	6.6979	Wt. dish + ash	23.4016
" "	6.6799	" "	23.4012
	.0580	Wt. ash	.0004

Dis. in N HCl , made alk w. N NaOH , gave CaCl_2 w. Na_2CO_3 .

Acid Equivalent (Goebel)

2 cc. sample. 5 cc. $\text{AgNO}_3 \frac{\text{N}}{10} \times 1.013$
 2 cc. of centrifgd. soln. for titrn.
 $\frac{2}{7} \times 25 \times 1.013 = \text{cc. } \frac{\text{N}}{10} \text{AgNO}_3 \text{ for sample} = 7.24$
 cc. KCN (F x 1.03) 6.50 6.50 6.70

Ash
0.7%

29.2 mg. & 108 cc.

Acid Equiv. 270

.54 cc. used up by 2.92 mg. subst.
 = .00108 cc. $\frac{\text{N}}{10}$

Jun 20 - Repetition of N

Wt. bot D + subst 7.4972 7.4606
 7.4606 7.4224

cc. $\frac{\text{N}}{70} \text{HCl}$ (x.995)
 " " NaOH (x1.015)

	.0366	.0382
	15.00	15.00
	5.45	5.175
	14.92	14.92
	5.53	5.25
	9.39	9.67
	.11	.11
	9.28	9.56

mg. N 1.86 1.91
 % N 5.08 5.00
 Ash-free 5.12 5.04

N =
5.1%
Ash-free

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Amino-N after different reaction periods.
 3 cc. soln. dild. to 6. 2 cc. samples for anal.
 Blanks.

Shaken 2 min, let stand 0.5 hr., shake 3 min. $KMnO_4$ 1.5 min.
 T Bas. Rdy. In other blanks see 37 B, Dr. Goebel's book.
 22° 764.4 0.21

Same as above, but let stand 1 hr.

21° 764 0.31
 22° 764 0.27

Detns. $\frac{1}{2}$ hr. -Bl. Factor
 23° 764 0.45 0.24 5640

1 hr. 24° 763.7 0.51 0.22 5610

$$\frac{.24 \times 5640 \times 100}{1} = 13536 \text{ } \mu\text{H}_2\text{N per 100 cc. orig. soln.}$$

$$9 \text{ in } 100 \text{ cc. } .5115 \text{ mg.}$$

$$NH_2N = 2.6\%$$

FEB 24 - In view of result obtained on hydr. of S.S. 39 w. 2N HNO_3 ,
 Acid Hydrolysis repeated w. N HNO_3 .
 3 cc. soln. dild. to 6, 6 cc. 2N HNO_3 added, for 4 hrs.

OPTICAL ROTATION
 l=1 Bl. +.04 .04 .01 .03 .03 Rdy. +.26 .25 .24 .26

$$[\alpha]_D = \frac{.22 \times 100 \times 8}{1} = +176^\circ$$

Redg. Sugars 2 cc. sample

Bl. 20.05 (Goebel)
15.60

27.2%

$$4.45 \times .957 \times .318 = 1.35 \text{ mg. glucose} = 0.68 \text{ mg. glucose}$$

B 1 hr. longer

l=1. $[\alpha]_D = \frac{.17 \times 100 \times 8}{1}$

Bl. = +.04 .05 .03
 Rdy. = +.21 .21 .22

$$= +136^\circ$$

(Goebel)

Bl. 20.05
16.35

$$3.70 \times .957 \times .318$$

Hydrolysis of SSSI

Nov 11

0.5 gms of SSSI dried to const. wt. was dissolved in 50 cc of 75% (by wt) H_2SO_4 . Solution was slow, taking about 24 hrs. Solution allowed to stand at room temp.

After 24 hrs 1 cc was removed diluted to 5 cc with water and optical rotation determined.

Time.	α	$[\alpha]_D$	
24 hrs.	+ .52	+ 260°	$.52 \times 250 = + 260^\circ$
48 hrs.	+ .38	+ 190°	.5 x 1 quite dark.
72 hrs	+ .13 (5 dentals)	+ 130°	$.13 \times 250 = 130$.5 x .5

250
 .52

 500
 12500

51130.00
 260.00

Adsorption - Type A undried Alumina - 00372

a - test was gotten in supernatants so the entire was centrifuged, washed once, & recentrifuged. The washings gone - test. The Alumina was shaken with 500 c.c. $\frac{1}{5}$ Na_2CO_3 for two hours, centrifuged, washed twice, & re-shaken with 500 c.c. more. The whole supernatants after the first and second extraction (which lasted over night) the Na_2CO_3 was neutralized with with HCl and the whole was conc in vacuo, dialyzed in parchment until nearly free of chlorides, conc. again and finally, after centrifuging away the small amount of Alumina which had in the meantime come out, placed in a corrosion bag and dialyzed until free entirely of chlorides, concentrated and ppt from 15 c.c. vol with 5 c.c. ^{50% HCl} HCl 1:1 at 0°. washed free of chlorides and with 50% HCl & then with acetone. Dried in vacuo. .36 gm, or 72% recovery.

12/1 The supernatants from a third extraction with $\frac{1}{5}$ Na_2CO_3 gone a test and were worked up exactly as above.

1/4/25 Adsorption on Type I S.S. 37AD

An experiment was tried in which 10 mg type I in 10 c.c. water were shaken with .1 g Alumina at pH 5. For a couple of ^{over night} hours, but was found after centrifuging that complete adsorption occurred. With a similar weight of Coolin at pH 8 - off with phenyl phthalimide, no great amount of adsorption took place because a very strong test for S was obtained. The adsorbed stuff was shaken for a period of two hours with about 20 c.c. of half normal sodium carbonate, a small portion removed, centrifuged & tested for S.

1/5/25 .25 gm Type I (prop 37) dissolved in water and diluted with .6 gm of Type A undried Alumina. Made up to pH 5. After shaking for 3 hrs no test for S specifically but after standing overnight specific test given so 4 g more of Alumina were added. This also failed to bring all S out. pH taken & found to be 6.5 so it was made up to 4. again & after an hour's shaking, no S could be detected.

1/7/25 The ppt (after cent), was ~~was~~ shaken overnight with 500 c.c. of $\frac{1}{5}$ normal Na_2CO_3 , cent & extracted twice with two lots of $\frac{1}{5}$ normal 250 c.c. carbonate. Each extraction lasting 4 hours. These last two fractions were kept separate from first one. The ~~old~~ Alumina was washed free of S & extracted overnight a fourth time with 500 c.c. $\frac{1}{5}$ Na_2CO_3

1/9 - The Alumina extract was neut, conc, & dialyzed in running water. Conc again and dialyzed in dist water till Cl free.

1/28. The rest from dialysis was conc to 15 c.c. vol in vacuo and the S cut off. The mother liquor contained a slight amount of S and was added to the 2nd & 3rd extract. The white ppt was washed with acetone & cent. repeated 3 times. Washed with acetone onto a paper & dried in vac. desiccator. Yield from first 2 gm .17 gm.

Biological Distribution 1:6, 1:10, 1:100

S.S. 37 B

Analysis of Propyl Sulfate after adsorption with alumina.
First extract.

Bottle #1 + Sample 6.7500
6.7496
6.7496

Diluted to 16.32 c.c. giving a 1:200 soln.
Spectrometer so it was centrifuged for 1/2 hrs
in case centrifuge to deposit residual
probably alumina, and the specific
rotation taken on this product.

" - Sample 6.6680
0.0816
Ash 64
0.0752

① Rotation: Zero +.05 Alpha and Zero 1.37

$\alpha = 1.32^\circ$

$[\alpha]_D = 264^\circ$
297

1.36
1.37
1.37

57
+287
ash-free

wt. bot + subst 6.6684
" " 6.6155
Ash .0529

wt. dish + ash 23.4058
" " 23.4016
.0042

Ash = 7.9%

② Ammo Nitrogen

	Blank	Temperature	Barometer (corrected)
	.14 cc.	22°	771.7
	.14 "	"	
	.13 "	"	
Run 10 min	.20	25.5	762
Blank + Ammo nitrogen from 2 cc of the above sample in a 1:400 soln	.34	22°	
	.34	"	"
Run 10 min	.48	25.5°	

c.c. Actual Ammo nitrogen = .20

Factor from table = .5730

$\frac{.20 \times .5730 \times 100}{1} = 11.46 \text{ mg. NH}_2 \text{ N per 100cc.} = 500 \text{ mg. subst.}$

2.3

NH₂N = 2.5%

$\frac{.28 \times .5550 \times 100}{1} = 15.54 \text{ mg. NH}_2 \text{ " " " "}$

3.1

" " = 3.4%

③ Silver Equivalent: 2 c.c. soln (1:200), 5 c.c. AgNO₃ (F=1.03) cont. and 2 c.c. for titration, or 2/7 of the original.

c.c. KCNS (N/50 F=1.03) =
Silver equivalent

1 6.45
2 6.95
230 Analyz = 6.50

Total nitrogen on 37 standard part. 2 c.c. of a 1:2000 color used for each sample

	1.	2.	Blank
c.c. N/50 (F=998) HCl	5.00	5.00	5.00
" N/100 (F=1.014) NaOH	6.70	6.60	9.30
	6.65		

Blank = .27 c.c. of N/50 Acid.

Sample = 1.62 c.c. of N/50 Acid

Difference = 1.35 c.c. N/50 and neutralized

$$\% \text{ Nitrogen} = \frac{1.35/50 \times 0.14}{.01 \times 92} = 4.10\% \text{ (ash free)}$$

Reducing sugars with Nitroson Acid. 1/2 c.c. of a 1:200 solution of 37 std. (2.5 mg) was pipetted into a sugar dish and 1 c.c. of v. slight NaNO₂ added + 4 c.c. glacial acetic acid. Allowed to stand (together with a blank) for 1/2 hr. Then 2 1/2 c.c. of 2N NaOH added and a blast of air blown through for 10 minutes. Heat with Na₂CO₃ + a regular sugar detn made on residue.

	1	2	Blank
c.c. Thiocyanate	15.35	15.05	15.20
Blank - " x F=957	= 4.65		20.05
Copper equivalent	11.48 mg		
Glucose "	73		
% Glucose	29.4%		

31.7%
 Ash free

SS 38

	Tag Nos	Tag dates	No. Flasks	Precipitation	Marked
10/24/24	1/193/11,12	\bar{X} -15, 16	5, 3	10/25	10/27
10/27/24	1/193/12	\bar{X} -16	14	10/28	10/29
10/28	1/193/11,12	\bar{X} -15, 16, 20	1, 3, 10	10/29	10/30
10/29	1/193/11	\bar{X} -20	14	10/30	10/31
10/30	1/193/11,12	\bar{X} -15, 20, 22	9, 4, 1	10/31	11/1
10/31	1/193/11,12	\bar{X} -20, 22	1, 14	11/1	11/3
11/5	1/193/194/12,5	\bar{X} -22, 28	4, 10 ⁹⁴	11/6	11/7
11/6	1/194/5	\bar{X} -28	14	11/7	11/8
11/7	1/194/5	\bar{X} -28	14	11/8	11/9
11/10	1/194/5,6	\bar{X} 28, 30	9, 5	11/11	11/12
11/11	194/5,6,7	\bar{X} 28, 30, \bar{X} 1-1	1, 12, 1	11/13	11/14
11/12	1/194/7	\bar{X} 1-1	14	11/13	11/14
11/18	1/194/192/7,4	\bar{X} 1-1, 4	2-12	11/19	11/20
11/19	1/192/194	\bar{X} 14, 12, 1	5, 3, 6 ¹⁰⁰	11/20	11/21
11/20	1/192/4,5	\bar{X} 1-12, 11	7, 7	11/21	11/22
11/21	1/192/4,5	\bar{X} 1-11, 12	4, 10	11/22	11/22

NOV 24 — At vol. of 4 l., 1:1 HCl stirred in to max.

pptn. Acid to Congo. Centfgd; ppt. washed w. $\frac{N}{40}$ HCl.

At vol. of 4700 cc. pptd. in cold w. 3 l. alc.

or let stand over night in ice-box

NOV 25 — Supernatant, —, siphoned off, ppt. centfgd. & washed 1" w. $\frac{N}{2}$

HOAc. NOV 26 — Susp. in H₂O, dissd. w. sl. xs. NaOH, centfgd. repeatedly in cold until clear. NOV 28-9 — Ppt. washed w. H₂O, 10 g. NaOAc added to soln.

at vol. of 700 cc., pptd. w. 350 cc. alc. Let stand in ice-box, DEC 1 —

Supernatant —, so centfgd. in 4 bottles. Ppt. in each taken up in 50 cc. $\frac{N}{2}$

219 flasks

328.5 l.

2

3

- HOAc. DEC 2 - Centfgd. in cold. Supernatant - . Ppt. taken up in H_2O , made faintly alk. w. 10% NaOH, centfgd. ~~John. sent. HOAc, 10g.~~ NaOAc added, & at vol. of 500 cc, pptd. w. 50 cc. alc. DEC 3 - Much S. in ppt, but also in filt. 50cc. more alc. added & set in ice box. After several hrs. portion centfgd. in cold. Test very weak. 25 cc. more alc. added. DEC 4 - Supernatant \pm , discarded, ppt. taken up in 250 cc. H_2O , centfgd. DEC 6 - 10g. NaOAc added, & at vol. of 400 cc, pptd. w. 125 cc. alc, set in ice-box. ^{Centfgd.} Supernatant, \pm , ~~sucked off, ppt. centfgd.~~ discarded. Ppt. taken up in H_2O . Made faintly alk. to effect complete soln., centfgd. DEC 10 - At vol. of 300 cc. (+10cc. NaOAc) pptd. w. 200 cc. alc. ^{Let stand in cold} DEC 11 - Supernatant -, centfgd. off. Ppt. dis. in H_2O , centfgd. DEC 16 - Poured from trace of ppt., & at vol. of ca. 200 cc, pptd. by adding $12\frac{1}{2}$ cc. HOAc & mixing thoroughly. Let stand in ice-box. DEC 17 - Centfgd. DEC 24 - Ppt. dis. w. acid xs. dil. NaOH, soln. neutr. w. $N HCl$, & 10g. NaOAc added. At vol. of 250 cc, pptd. w. 150 cc. alc, let stand in cold. DEC 26 - Supernatant \pm , so mixt centfgd. in cold. DEC 29 - Supernatant poured off, ppt. rediss. in H_2O . ^{Centfgd.} JAN 8 1925 - Made acid to Congo w. 1:1 HCl, & at vol. of 250 cc. pptd. w. 700 cc. alc. ^{chilled} After several hrs. in ice-box, ^{sup. amt.} supernatant, weakly \pm , sucked off, remainder centfgd. in cold. Supernatant (X) + ca. = vol. Et_2O gave slight ppt. Main ppt. dis. in H_2O + little $N HCl$, but soln. thick & opalescent, so NaOH in xs. added. Neutr.
- JAN 9 - Centfgd., 5g. NaOAc & 10cc. $N NaOH$ added. At vol. of 200 cc, pptd. w. 250 cc. alc, let stand in cold. JAN 10 - Taken up in H_2O & pptn. repeated under exactly the same conditions, as most of color had been removed by previous pptn. Supernatant \pm , discarded, ppt. centfgd. Ppt. from X above dis. in H_2O , soln. $++$, neutr., added to (39) JAN 12 - Main ppt. dis. in H_2O , centfgd. JAN 13 - Neutr. w. 25% HOAc, & at vol. of 200 cc, 12 cc. HOAc added. Thoroughly stirred. Centfgd. JAN 15 - Supernatant poured off, ppt. stirred up w. H_2O , acidif. to Congo w. 1:1 HCl, & at vol. of 225 cc, practically colorless soln. pptd. w. 600 cc. chilled redistd. alc. After several hrs, supernatant -, removed by centfg. Ppt. taken up in H_2O . 16" x 17" - transf. to dialysis bags, rinsed in w. little $N HCl$, dialyzed agst. running H_2O for several hrs, then ~~agst.~~ in ice-box agst. successive changes distd. H_2O until Cl-free inside & out. Let stand in ice-box under H_2O + acetone. FEB 19 - Washed w. acetone, filtered, dried. Yield: 2.7g.
- MAR 2 - Since analytical values not quite O.K., 2.6g. diss.

S.S.S. 38

FEB 20 - Wt. subst. + bot(11) dried abt. 6. CHCl₃ 8.6480 8.6418 864.14 8.6414

Wt. bot + subst. 8.6415
 " " 8.5390
 " " .1025
 Ash .0004
 .1021

Diss. in H₂O + little N H₂SO₄, N NaOH added until ppt. formed rediss. made up to 20.42 cc. Soln. centrifgd. from trace suspended matter.

Wt. bot + subst 8.5390
 " " 8.4636
 .0754

ASH

Wt. Pt dish + ash 23.4010
 " " - 23.4007
 .0003

Ash
0.4%

(Wt. bot. + mult 8.4637
 " " 8.4051
 .0586

Wt. Pt dish + ash sulfate 23.4008
 " " " 23.4005
 .0003

FEB 26

OPTICAL ROTATION

l=1 d = -0.02 -0.03 -0.03 Rdg.: +1.44 1.43 1.44

[α]_D = +294°
Ash-free

MICROKJELDAHLS

2 cc. samples

cc. N/50 HCl (x)
 " N/100 NaOH (x 1.014)

I	II
3.00	3.30
15.79	10.71
<u>12.81</u>	<u>8.00</u>
2.98	2.71

b. 3 hrs.

HYDROLYSIS w. b. N HNO₃

2 cc. dil. w. 2 H₂O, 4 cc. 2N HNO₃

Rotation: l=1 d = -0.07 -07 Rdg.: +19.19 [α]_D = +208
 Redg. Sugars: 2 cc. sample (Goebel) Blank 20.05
 Rdg. 17.90

b. 1 hr. longer

Rotation: l=1 d = +0.04 .04 Rdg.: +28.32 .32 .33 [α]_D = +224
 Redg. Sugars: 2 cc. sample (Goebel) Blank 20.05
 Rdg. 17.45

18.4%

2.60 x .318 x .957 = .796 = .46

Test for Ammonium Salts

2 cc. dil. to ca. 5 cc., 5 cc. 10% aq. Na₂CO₃ added. Aspirated.

cc. N/50 HCl (x)
 " N/100 NaOH (x 1.014)

2.00
 12.90
9.00
 3.90

FEB 26 -

Stability to approx. 5% NaOH at room temp.
 2 cc. soln. lev stand + 1 cc. H₂O + 3 cc. 10% NaOH over night.

Stability to htg. w. approx $\frac{N}{10}$ NaOH.
 2 cc. soln. + 3 cc. H₂O + 5 cc. $\frac{N}{5}$ NaOH

cc. $\frac{N}{50}$ HCl (r))	2.00
" $\frac{N}{100}$ NaOH (x1.014))	14.61
		10.71
		<hr/> 3.90

Hydrolysis w. HNO₂

4 cc. soln. + 6 cc. 30% NaNO₂ + 3.5 cc. HOAc lev stand ca. 0.5 hr.
 1.2 cc. 18N H₂SO₄ added & HNO₂ aspirated out for 0.5 hr. FEB 27 - Dild. to
 20 cc.

ROTATION

$l=2$ $\alpha = -0.03$ $\alpha = -0.02$ $\alpha = -0.02$ Rdy.: +0.32 .33.32

$[\alpha]_D = \frac{+34 \times 10}{2} = +170^\circ$

Redg. Sugars

cc. Na₂S₂O₃ (~~soln.~~)

3 cc. sample
 Blank (Goebel) 20.05
 Rdy. 16.60

$3.45 \times 957 \times 3.18$
 $= 1.05 \text{ mg. Cu} = .56$
 $> 18.7\%$

S.S.S. 38 A

Ba(OH)₂
MAR

w. xs. dil. HCl, made faintly alk. to litmus w. NaOH, pptd. w. xs. aeth.
 Ba(OH)₂ (Kahlbaum's for analysis) soln. Let stand in ice-box over night.
 MAR 3 - Centfgd. repeatedly. Opalescent supernatant still +++ . Addn.
 of even enough Ba(OH)₂ to ppt. out on cooling does not ppt. all S. MAR 23 -
 Ppt. (A) taken up in sl. xs. H₂SO₄. Supernatant B^{old} pptd. w.
 xs. H₂SO₄, centfgd., neut. w. NaOH, concd. to dryness in vac.
 taken up in washings of H₂SO₄ ppt., again neut., & at vol. of
 ca. 150cc., pptd w. 100cc. alc. MAR 24 - Supernatant discarded
 ppt. centfgd., stirred w. H₂O, set in ice-box. (A) repeatedly centfgd.,
 ppt. washed w. little dil. HCl & centfgd. Solns. ^{made pl. alk.} w. aq. NH₃.

B lost in accident to cold centrifuge.

APR 2 - A solns. + 10g. NaOAc, at vol. of 250cc, pptd. w.
 200cc. alc. Let stand in cold. APR 6 - Supernatant -, mixed.
 centfgd. APR 7 - Ppt. diss. in ca 150cc. H₂O & milky soln. pptd. by
 adding HOAc to neutrality, then 4cc. more. Stirred thoroughly,
 let stand in cold. APR 8 - Centfgd, APR 20 - Diss in H₂O + sl. xs.
 dil. NaOH, dild. to 250cc. & run through Berkefeld N. Still opales-
 cent, so run through Berkefeld W, now coming through practically
 clear. Washings opalescent, so added to (44). APR 21 - Concd. in vac.
 APR 22 - At vol. of ca 150cc., again pptd. w. HOAc, considerable
 xs. being necessary. Centfgd. Supernatant, -, discarded.
 Ppt. taken up in dil. HCl, pptd. w. a 1st vol. of ca. 100cc., pptd.
 w. 150cc. ^{redist.} alc. Let stand in cold 1/2 hr., centfgd. Sm. amt.
 active material pptd. from supernatant w. Et₂O, added to main
 portion. Diss. in ca 75cc. H₂O + HCl, repptd. w. redistd. alc.
 Let stand 2 hrs., centfgd., ppt. taken up in H₂O, repptd. w. ca.
 3 vols. redistd. alc. Let stand in cold, centfgd. Ppt. taken
 up in H₂O, not diss. this time. Let stand over night. MAY 5 - xs. redistd
 alc. added, let stand, centfgd. in cold. Repeated 1st, then let stand
 again w. H₂O. MAY 6th Centfgd., ppt. taken up in H₂O, centfgd. again
 after several hrs. Repeated. Ppt. again taken up in H₂O, let stand
 over night. Centfgd. Repeated. Now Cl⁻ free, so taken up 2x in
 acetone, then filtered, sucked off, air-dried in warm place, then
 in vac. over night. Yield: 2.0g.

S.S. 38 A

MAY 11 - Wt. bot (4) + subst. 8.3943 8.3904 8.3886 8.3876 8.3869
8.2305

MAY 27 - Diss. w. dil. HCl + NaOH, made up to 30.72 cc; centrifgd. Corr. f. Ash .1564
.0028
.1536

MAY 12 ASH

Wt. bot + subst 8.2307
" bot 8.1534
.0773

Wt. dish + ash 23.3910
" dish 23.3909
23.3895
.0014

Ash
1.8%

OPTICAL ROTATION

l = 1. 0 = 0 +0.01 -0.01 Redg. = +1.54 1.55 1.55

$[\alpha]_D = 155 \times 2 = +310^\circ$

$[\alpha]_D = +310^\circ$
(ash-free)

Microkjeldahls 2 cc. samples
cc. N/70 HCl (x0.995) 10.00 9.95
" NaOH (x1.015) 7.16 7.32
7.18 2.63
7.26 .15

$2.48 \times 2 = 4.96 \text{ mg. N}$
 $= 0.5 \text{ mg. N}$

Amino-N (Van Slyke)

JUN 2 - Bar. 761.5 Temp. 29°

Redg. Bl. 0.23 0.20
0.69 Mean: .215
.69
.205

N = 50%
Ash-free

$\frac{.495 \times .544 \times 100}{2} = 12.9 \text{ mg. NH}_2\text{N per 100 cc. soln.}$

NH₂ N
= 2.6%
Ash-free

Action of HNO₂

JUN 1 - 2 cc. soln. + 4 cc. 30% NaNO₂ + 2.2 cc. HOAc let stand 0.5 hr. cooled w. ice-H₂O, 0.8 cc. 0.8N H₂SO₄ added, aspirated 1 3/4 hrs. made up to 25 cc. 5 cc. samples

Bl. Redg. Sugars 19.30 (Gebel) 16.25 (overrun) 16.30 Mean: 16.25
16.25

$3.05 \times 1.01 \times 3.18 = .98 \text{ mg. Cu} = 0.57 \text{ mg. glucose} = 28.5\% \text{ glucose}$

28.5% glucose

Rotation

l = 2 0 = -0.01 0 +0.01 Redg. : +.18 .15 .16 .18 .16 .17

$[\alpha]_D = \frac{.17 \times 150}{.04 \times 2} = +213^\circ$

bl 100
17

S.S. 38A

JUN 2
JUN 3

HYDROLYSIS w. 1.4N HNO₃:
2cc. dil. to 4, 4cc. 2N HNO₃ added, for 5 hrs. Left over ab. in cold.
~~2cc. sample~~ conty. w. NaHCO₃, made up to 10cc., 2cc. taken

for anal.
Bl.

19.30 (Goebel.) Redgs: 14.70 14.70

14.70

$4.60 \times 1.01 \times 3.18 = 1.48 \text{ mg. Cu} = 0.77 \text{ mg. glucose} =$

38.5% Redg. Sugars

~~wt. bot.~~

Hydrolysis w. 1/2 H₂SO₄
wt. bot. + subst

8.1538

8.1260

.0278

?

Jun 23 Repeated by Goebel.

4cc. dil. to 8, 8cc. 3N HNO₃ added.

Time:	4 hrs	5 hrs.	6 hrs.
Sample	2cc.	2cc.	2cc.
Na ₂ S ₂ O ₃ (8/1.01)	16.15	15.50	15.20
Cy Eq.	1.03	1.26	1.35
Glucose	.59	.68	.75
% "	23.6	27.2	28.6

Bl. 19.40

28.6%
Redg. Sugars

SEP 28 1925 — 1 g. 38A (contg. few mg. left of 37) diss. w. shaking, in 5 cc. conc. HCl, cooling w. ice-H₂O. Material gelatinized & diss. in viscous soln. very slowly, so 5 cc. more conc. HCl added. SEP 29 — Still + + + +, so removed from ice-box & let stand at room temp. SEP 30 — A.M. + P.M. ±
 OCT 1 — Now —. 0.1 cc. withdrawn for redg. mg. detn. 21%. Remainder centfgd. from trace amorphous matter, & clear supernatant concd. in vac. at bath temp. not exceeding 46.5°, taking care not to let concentrate get that hot. Finished on High vac. pump until very viscous syrup. Ca 25 cc. abs. alc. (redistd.) added. OCT 2^{1st} —
 Worked up w. abs. alc. until solid, centfgd. & ppt. washed 2X w. sm. amt. abs. alc., then acetone, & dried: (A). 0.1062 g.
 Supernatant + alc. washings (B) concd. At about $\frac{1}{2}$ vol. more of insol. subst. sepd. This centfgd. off, washed, w. a little abs. alc., ~~then~~ w. acetone dried, added to (A). Soln. (B) concd. in vac. to thick syrup, part of which solidified (apparently amorphous, but became syrupy again on standing. Rediss. in abs. alc. & let stand in ice-box. Remained clear, so concd. dry, finally in high vac. OCT 22 — Taken up in 5.5 cc. abs. alc., 5.5 cc. dry Et₂O cautiously added, w. stirring. Ppt. centfgd. off, washed 2X w. 3 cc. ea. time abs. alc. - Et₂O, taken up in redistd. acetone filtered off, washed. Ppt. (B): 0.44 g. Clear reddish M.L. & washings let stand in \equiv ice-box. OCT 27 — Nothing had sepd., so concd. dry in vac., finally in high vacuum. Taken up in 10-12 cc. H₂O, centfgd. from sm. amt. dark brown amorphous material, & lt. brown soln. dild. to 25 cc. (C).

A - gives weak Cl⁻ test, strong Molisch. After continued drying over CaCl₂ + NaOH — practically Cl⁻ free

B - contains much Cl⁻.

SSS 38A

Fraction (B)

NOV 18 — 0.33 g. B diss. in 5 cc. H₂O (contg. 26.7 mg. Cl = 27.5 mg. HCl)
& treated w. 0.12 g. AgNO₃, as to the HCl in the sample, + 0.18 g. in xs., adding
1.2 cc. N HCl & to the xs. AgNO₃. After 4 hrs. 0.3 g. more AgNO₃ &
2 cc. N HCl (slight deficiency) added. 19" ^{off} AgCl filtered off, washed
well w. H₂O, & H₂S passed into filtrate to remove xs. Ag. Ag₂S
centrifgd. off, soln. coned. to dryness in vac, finally w. abs. alc.,
taken up w. little abs. alc., let stand in ice-box. No repr. of crystals.
Dry Et₂O added until faint oily ppt., & mixt. seeded w. crystals of
2,5-anhydroglucose, but none reprecipitated.

This fraction gives strong naphthoresorcin test.

S.S.S. 38A

Hydrolysis

OCT 7	-	Fraction A	wt. bot. + desicc. dried subst.	7.3542
			"	7.2480
		Dried in vac at room temp. over CaCl ₂ + KOH	7.3507	.1062
			7.3506	
OCT 20		Diss. in H ₂ O made up to 10 cc.	7.3006	
			.0500	

OPTICAL ROTATION

l=1 Bl. = -0.06 05 07.06 Rds: +0.96 .98 .98 .99

$[\alpha]_D + 2.08^\circ$

MICROKJELDAHLS

cc. N/70 HCl	10.00	10.00	Bl	10.00	Bl	10.00
" " NaOH (x.934)	8.38	8.40	10.60	10.00	10.60	
		7.84		9.90		
		2.16		.10		

$2.06 \times 0.2 = 0.412 \text{ mg N}$

N Total
 =
 4.1%

Reducing Sugars 1cc. samples

Blanks 18.93
 14.63
 4.30

Titrs. 14.65 14.60

$4.30 \times 1.025 \times .318 = 1.40 \text{ mg. Cu} \approx 0.69 \text{ mg. glucose}$

Redy. sugars = 13.8%

OCT 23

Amino N

Barom	761	Bl.	0.15	Det.	0.58
T	22.5°		.13		.14
	21.5°				.44

$.44 \times .5643 \times 100 = 12.4 \text{ mg. NH}_2 \text{ N per 100 cc.}$

$\text{NH}_2 \text{ N} = 2.5\%$

OCT 28

Hydrolysis

wt. bot. + subst. 7.3010
 7.2935
 .0075
 6 cc. N/2 HCl added
 b. 2 hrs.

Dil. to 7.5 cc., 2 cc. samples taken

NH₂ N
 =
 2.5

Blank 18.93
 Detn. 17.70

$1.23 \times 1.025 \times 0.318 = 0.40 \text{ mg. Cu} = 0.29 \text{ mg. glucose}$

1 cc. soln. after 2 hrs. longer in sealed tube in b. H₂O
 Redy. sugars 14.5%

Blank (Sebel) 1cc. conc. HCl 19.00
 17.85
 w. 1 cc. dil. HCl 19.00

$.65 \times 2.126 \text{ Cu} = 21\% \text{ glucose}$

$1.15 \times 1.025 \times .318 = .375 \text{ Cu} = .28 \text{ glucose}$

2.2.2.38A

Hydrolysis

Fraction A

part. part. base, mixed alkyl.

Blank in case of room temp. over CaCl₂ + KOH
1.3207
1.3206
1.3006
0.0200

OPTICAL ROTATION
R = -0.02 0.50 0.02 0.50 0.02 0.50 0.02 0.50 0.02 0.50 0.02 0.50

$$[\alpha]_D^{20} + 5.08$$

MICROKJELDAHL'S

10.00 10.00 10.00 10.00 10.00 10.00 10.00 10.00
8.40 10.00 10.00 10.00 10.00 10.00 10.00 10.00
1.84 1.84 1.84 1.84 1.84 1.84 1.84 1.84
2.16 2.16 2.16 2.16 2.16 2.16 2.16 2.16
10.00 10.00 10.00 10.00 10.00 10.00 10.00 10.00
10.00 10.00 10.00 10.00 10.00 10.00 10.00 10.00

Total N = 1.1%

$$1.0 - 5.06 \times 0.2 = 0.41504$$

Reducing Sugars 1cc. samples

Titres. 14.63 14.60

14.63
18.93
4.30 x 1.025 x 3.18 = 1.4207 cc. x 0.2000 glucose
1.4207 x 0.2000 = 0.28414 g.

Amino N

Redg. Sugars after HNO₂ Treatment.

2 cc. + 4 cc. 30% NaNO₂ + 2.2 cc. HOAc. after 1/2 hr. 1.5 cc.
30% (vol.) H₂SO₄ added, aspirated 1 1/2 hrs. Dild. w. H₂O, made
up to 25 cc., 5 cc. samples taken.

Blank: 18.93
Dets: 15.30 15.30

$$3.63 \times 1.025 \times 3.18 = 1.18 \text{ ang. Cu} = 0.6 \text{ ang. glucose}$$

Redg. Power
30%

Blank 18.93
Dets 15.30

1 cc. of the above in each tube in 1.5 cc. H₂O
1.53 x 1.025 x 3.18 = 0.4995 cc. x 0.2000 glucose
0.4995 x 0.2000 = 0.0999 g.

Blank (total) 18.93
Dets 15.30
1.12 x 1.025 x 3.18 = 3.63
3.63 x 0.2000 = 0.726 g. glucose
0.726 x 0.2000 = 0.1452 g. glucose

S. S. S. 38A

Hydrolysis Fract. B

OCT 23 - Portion dried in vac. at room temp. over H₂SO₄

+ NaOH in bot(12) 8.1698 8.1686
 8.0676
 .1010

Diss. in H₂O, made up to 20.2 cc.

OCT 22/6

OPTICAL ROTATION

P = 1 Bl. -0.06 .06 .06 Rdgs. +0.85 .88 .87 .89

$[\alpha]_D = +93 \times 2 = +186^\circ$

$[\alpha]_D = +186^\circ$

MICROKJELDAHLS 3 cc. samples

cc. N/70 HCl 10.00 10.00 10.00
 " " NaOH(+9.34) 7.20 7.20 6.72
 3.28
 Bl. 10

Total N
 4.2%

$3.18 \times 0.2 = .636 \text{ mg. N in } 15 \text{ mg.}$

Amino N

Barom 757 Setu: 0.50
 T. 22° 22° 0.95
 Blanks: 0.095 - Bl.
 .085
 .105
 $.405 \times 5613 \times 100 = 11.4 \text{ mg. N in } 100 \text{ cc.}$

NH₂N
 2.3%

Chlorides

cc. HNO₃-AgNO₃ N/20 3.00 1cc. sample 3.00 2cc. sample
 " " " N/50 7.50 7.50
 " N/50 KSCN(+1.084) 6.36 6.89 5.87 6.36
 .61
 $1.14 \times .709 = .808 \text{ mg. Cl in } 10 \text{ mg.}$

Cl =
 8.1%

Redg. Sugars as Glucose

1 cc. samples
 Blank 18.93 18.93
 Rdgs: 15.55 15.35
 $3.58 \times 1.025 \times .318 = 1.17 \text{ mg. Cu} = 0.6 \text{ mg. Glucose}$

Redg Power
 12%

2 cc. samples

Hydrolysis - 2 cc. dil & 5 5cc. 4N HCl added.

B 2 hrs: Blank: 18.93 18.93 Mean Redg. Power
 Titm: 15.33 15.38 15.36 30%

B 4 1/2 hrs. Same as above: Glucose = 0.6 mg.
 Blank (Gebel) 19.00 19.00 19.00
 15.65 15.75 15.70

$3.30 \times 1.025 \times 0.318 = 1.08 \text{ Cu} = 0.56 \text{ mg.}$
 28%

S.S.S. 38A

Titration of AgNO_2 .

NOV 17 - .1000 g. diss. in H_2O , little N HNO_3 added, $\text{K}_2\text{Cr}_2\text{O}_7$ cooled, made up
 to 100 cc. 25 cc. samples
 cc. NH_4SCN ($\times 1.001$) (1.93) 1.95 1.96 1.95 cc 9.75 mg. NaCl
 6.085 " HCl
 .100 g. \propto 0.0243 g. HCl \propto 0.685 cc. N HCl

S. S. S. 38 A Hydrolysis

Fract. C.

OCT 27 - 0.5 cc. main soln. dild. to 5, 1cc. taken ea. for sugar anal.
 Blank: 18.93
 Titcn: 18.95

OCT 28 - 0.5 cc. main soln. used

Bl.	18.93		18.93
	15.94	15.86	15.90

$$3.03 \times 1.025 \times 3.18 = 0.99 \text{ mg. Cu} = 0.52 \text{ mg. glucose in } 0.5 \text{ cc.}$$

$$26 \text{ mg. " in } 2.5 \text{ cc.}$$

OCT 30

Hydroly. w. Stronger Acid.

~~Potts~~ 2cc. dild. to 4 w. 4N HCl. B. 2 hrs. 1.5 cc. samples
 Blanks (Goebel) 19.00
 Rds. 4.25 4.50

OCT 31

1 cc. hydrolyzed material dild. filtered, washed to 5 cc.

Blank: 18.93
 Rds: 18.00 18.00

1 cc. samples. = 0.1 cc. orig.

$$\therefore \text{ in orig soln } 250 \times 0.25 = 62.5 \text{ mg. glucose if hydroly. further}$$

$$0.93 \times 1.025 \times 3.18 = .3 \text{ mg Cu} = 0.25 \text{ mg. glucose in } 1 \text{ cc.}$$

OCT 30

Microkj.

1/2 cc. samples

cc. N/70 HCl	10.00	10.00
" " NaOH (x.934)	9.40	9.35

Wt. of fraction.

2 cc. used

Est. dish + residue: 14.3564 14.3555

" " 14.3358 14.3358

$$.0197 \times 2.5$$

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	No. on FLA.	Date on FL	No. of FL	Planip	Stacked
11/24/24	1/192/5	XI-12	14	11/20	11/26
11/25	1/192/193/5,6	XI-12,19	5,9	11/26	11/28
11/26	1/193/6	XI-19	14	11/28	11/29
12/1	1/193,195/6,3	XI-19,20	7,7	12/2	12/3
12/2	1/195/3	XI-20	14	12/3	12/4
12/3	1/193,195/5,3,4	XI-18,20,22	1,9,4	84	12/4
12/4	1/195/3,4	XI-20,22	1,13		12/5
12/5	1/195,196/4,3	XI-22,29	3,11		12/6
12/8	1/196/3	XI-29	14		12/9
12/9	1/196/3	XI-29	11	137	12/10
12/12	1/196/4	XII-6	14		12/13
12/15	1/196/5	XII-8	14		12/16
12/16	1/196/4,5	XII-6,8	4,9		12/17
12/17	1/196/5	XII-8	14		12/18
12/18	1/196/5	XII-8	13	205	12/19

DEC 22 - At vol. of 4200 cc., pptd. w. 1:1 HCl to max. turb. Soln. not yet acid to Congo. After ca. 2 hrs. ppt. centrifgd. off, washed

1" w. H₂O slightly acidulated with HCl. At vol. of 4700 cc., chilled soln. pptd.

w. 3000 cc. ~~old~~ cold alc. DEC 23 - Supernatant -, centrifgd. off

(Contains considerable active material - X) in cold, ppt. washed w. 1/2 N.O.A.C. DEC 24 - Relatively slight amt. ppt., after removal of - supernatant, homogenized w. H₂O. DEC 26 - Centfgd. Somewhat colored supernatant, ±, discarded, ppt. again mixed w. H₂O.

DEC 29^{ref.} - Diss. w. 2.2% NaOH. X in meantime dissolved again washed w. H₂O & solvent, added to main soln. Ppt. diss. w. acid of NaOH & soln.

307.5
liters

168
37
205

2

S.S.S. 39

+, pptd. w. alc. Ppt. rediss. as before, repptd. w. alc. Ppt. taken up in H₂O, acidified to Congo w. 1:1 HCl, centrifgd. Supernatant, now pale brown, neutralized, ++. Added to main portion. JAN 7 1925 - At

3 vol. of 500 cc., 20 g. NaOAc added, pptd. w. 250 cc. alc, set in cold. JAN 8 - Supernatant, -, mixt. centrifgd. JAN 9 - Ppt. taken up in H₂O, centrifgd. JAN 10 - 15g NaOAc added to clear soln, & at vol. of 450 cc., pptd. w. 250 cc. alc. JAN 12 - Supernatant -, mixt. centrifgd. JAN 13 - Ppt. taken up in 250 cc. H₂O, centrifgd. Clear soln. poured into 2 centfg. HOAc bottles, mixed into ea. w. ca 25 cc. H₂O, & each then acidified w. 8 cc. HOAc. Thoroughly stirred, centrifgd. Ppt. taken up in H₂O, neut. w. NaOH, centrifgd. Soln. + 10 g. NaOAc, at vol. of 200 cc., pptd. w. 125 cc. alc. let stand in cold. JAN 16 - Centfgd., supernatant, -, discarded. HOAc Ppt. taken up w. ca 300 cc. $\frac{1}{2}$ HOAc. JAN 17 - Centfgd. Ppt. taken up in H₂O, neut. w. NaOH. JAN 19 - 5g NaOAc & 10 cc. N NaOH added & at vol. of 200 cc., pptd. w. 250 cc. alc. JAN 20 - ^{Ppt. rediss. in H₂O & repptd. as before.} Supernatant +, 50 cc. more alc. added. 6 NaOH NaOH JAN 22 - Ppt centrifgd. off. Supernatant still +, so pptd. w. 2 vols. alc. & let stand in ice-box. Both ppts combined, dis. in H₂O, centrifgd. from dirt. HOAc JAN 26 - Soln., at vol. of ca 200 cc., acidif. w. 12 cc. HOAc & thoroughly mixed. Centfgd. after several hrs. JAN 27 - ~~Soln.~~ Ppt. dis. in H₂O + enough NaOH to make soln. slightly alk. Ba(OH)₂ soln. added. No ppt. at 1", but w. large excess considerable ppt. Ba(OH)₂ formed. Let stand in ice-box. JAN 28 - Centfgd. Milky supernatant (A) poured from chiefly gummy ppt. (B) and run through Berkefeld N 5 in candle, coming through clear. B washed w. H₂O contg. Ba(OH)₂ & washings also run through the filter. JAN 30 - Tetrate -. Ppt. washed w. ca. $\frac{1}{2}$ satd. Ba(OH)₂ & this run through the filter. FEB 3rd - Washings also -. Filter washed through repeatedly w. very dil. HCl. Main ppt. B also dis. by shaking w. xs dil. HCl & Ba⁺⁺ removed from both portions by pptng. w. xs H₂SO₄. Centfgd, neut., con'd. in vac. FEB 5 - At vol. of ca. 150 cc., pptd. w. 100 cc. alc. let stand in cold. FEB 6 - Supernatant -, centrifgd. off. Ppt. taken up in H₂O, centrifgd. FEB 7th - Soln. acidified w. 1:1 HCl, ^{at vol. ca 125 cc.} pptd. w. 100 cc. cold alc. + 200 cc. abs. alc. redistd. Let stand in ice-box $\frac{3}{4}$ hr., centrifgd. Supernatant -, ppt. dis. in H₂O soln. diald. 1" agst. running H₂O, then agst. distd. H₂O in ice-box until free from Cl⁻ & SO₄²⁻. Yield: 0.8g. + contents of bags centrifgd; washed 2x w. redistd. acetone, ppt. filtered off, washed w. redistd. acetone, dried. Yield: 0.8g. + 0.6g. = 1.4g.

S. S. S. 39

FEB 14 Wt. bot (ii) + subst 8.5573 8.5524 8.5499 8.5469 8.5455 8.5447
 8.5444 8.5437 8.5433

FEB 18

Dis. w. 2% H₂SO₄, 1/2 N₂O₄ added until ppt. just rediss, make up to 22.0 cc.
 Ash Dctn. .1096 23.4018
 Wt. Pt dish + ash 23.4018
 " " " 23.4008
 " " " .0614 .0010

Ash dis. in drop HCl w. greenish color. Soln. gives strong Ca⁺⁺ test w. C₂O₄²⁻, showing ash mainly CaO.

Ash
1.6%

OPTICAL ROTATION

l = 1 0 = +0.02 .02
 Rds: +1.56 1.54 1.52 1.53

$[\alpha]_D = +304^\circ$

$[\alpha]_D +304^\circ$
Ash-free

MICROKJELDAHLS

cc. N/50 HCl (.998)	I	II		1.5 cc. samples	
		Bl. I	Bl. II	Bl. I	Bl. II
cc. N/100 NaOH (1.014)	2.00	2.00	2.00	2.00	2.00
	12.09	13.06	11.36	14.80	
	11.01	12.09	8.00	8.99	11.42
	1.08	1.04	3.36	3.42	3.38
Mean: 1.103	2.95	.97		.57	

2.38 x 0.14 = 0.333 mg. N in 1.5 cc. or in ^{7.5} 15 mg.

Amino N

FEB 19 - 1:400 Soln S.S.S. 39 Shaken 5 min.
 Blanks: 0.2 20°, 769, 0.2 21°, 769.
 Rds.: 0.43 21.5°, 0.42 21.°
 Factor .5733 Rdg - bl. = 0.22 cc.

N =
4.4%
Ash-free

$\frac{.22 \times .5733 \times 100}{2} = 6.3 \text{ mg. NH}_2 \text{ N per 100 cc. 1:400 soln.}$
 In 100 cc. 250mg.

HYDROLYSIS w. 2N HNO₃

2 cc. dil. & 4, 4cc. 4N HNO₃ added, b. 5 hrs.
 made up to 10 cc. 2 cc. for redg. sugars
 Bl. 20.05
 Rdg 19.35

NH₂ N
2.5%
Ash-free

$.70 \times .957 \times .318 = .213 \text{ mg. Cu} = .24 \text{ mg glucose} = 12\% \text{ Redg. Sug.}$

OPTICAL ROTATION

l = 1 0 = -0.04 .02 .02 Rds = .02 .01
 B. 2 hrs. longer Still optically inactive.

S.S.S. 39

FEB 20 - 4 cc. main soln. + 7cc. 30% aq. NaNO₂ + 3.7cc. HOAc let stand 0.5 hr. w. occasional shaking. 1.4 cc. 18N H₂SO₄ added & HNO₂ aspirated out for 1/2 hr. Rinsed into flask & made up to 25 cc.

OPTICAL ROTATION

l = 2 0 = -0.04 -0.03 -0.03 Rds. + 0.25 .27 .26

$$[\alpha]_D = \frac{.29 \times 100}{2 \times .08} = +181^\circ$$

Redg. Sugars 3cc. samples
Bl. 20.05 Rds. 15.65 15.90
15.80

4.25 x .957 x .318 = 1.29 mg. Cu = 0.65 mg. glucose

Redg. Sugars = 27.1%

3 x $\frac{20}{25}$ mg. = 2.4 mg.

G. 2 hrs.

ROTATION

l = 2 Bl. ~~.04 .04 .01 .03 .03~~ Rds. ~~.26 .25 .29 .26~~
Bl. .04 Rds. +.18 $[\alpha]_D = \frac{.22 \times 100}{2 \times .08} = +138^\circ$
.03 .18 .19 (Goebel)
.04 18 .18
.03 18 Redg. Sugars 3cc. sample

Bl. 20.05 (Goebel)
17.30
2.75

18%

$\frac{11}{17} \times 27$
 $\frac{17}{17} \times \frac{27}{17}$
 $\frac{127}{119}$

α N/50 HCl (x 998) 2.00
" N/100 NaOH (x 1.014)
 $\frac{7}{25} \times 20$ mg. 5.6 mg. subst.

NH₃ N

13.61
10.21
3.30

7cc. sample ^{centr.} & aspirated 1 hr. w. 10cc. satd. K₂CO₃.

4.00
3.35

.65 x .14 = .09 mg. N in 7cc.

NH₃ N =

1.6%

Acid Equivalent (Goebel)

2 cc. sample

5cc. AgNO₃ N/10 x 1.013

2 cc. centrifgd. soln. for titry

$\frac{2}{7} \times 25 \times 1.013 = \alpha$ N/50 AgNO₃ for sample = 7.24

α " KSCN (x 1.03)

6.80

6.60 6.60

.44 cc. used up by 2.86 mg. subst.

.0088 α N

28.6 g. = 88 α N

Acid Equiv
325

NH₃ N

8cc. sample centr.
w. NaOH, aspirated 1/2 hrs.
w. 10cc. 40% NaOH

α N/50 HCl (x 998) 2.50
" N/100 NaOH (x 1.014) 15.09
11.99
3.10

4.00
3.14

.86 x .14 = .12 mg. N

NH₃ N =

In sample $\frac{8}{25} \times 20$ mg = 6.4 mg.

1.9%

S. S. S. 39

- MAR ³ 2 1925 - 1.5 g. air-dry S.S. 39 diss. in 200 cc. hot $N HNO_3$. Turned pinkish at 1". Soln. b. under rf. for 4 hrs., becoming bright yellow, then greenish yellow. Cond. in vac., finally in warm desiccator to ca. 10 cc. 4 cc. conc. HNO_3 added & let stand over night.
- MAR 4 - Few glistening, rhombic crystals on bottom. Inst. b. few min. then stirred on watch-glass over b. H_2O until dry. Few cc. more 1:1 HNO_3 added, stirred again over b. HNO_3 . White residue taken up in acetone, crystn. residue remaining insol. Let stand ~~several hrs. @ room temp.~~ ^{over night.} ~~then in ice-box over night.~~ MAR 5 - Acetone had evapd. off. Residue taken up w. more acetone, filtered, washed with several sm. portions acetone. Filtrate (B). Ppt. contained much ash, app. Ca_2O_4 , so taken up in 5 cc. $N NH_4OH$, filtered, washed w. 2 cc. $\frac{N}{5} NH_4OH$, 0.3 cc. conc. HNO_3 added. Free acid rapidly sep. as prisms. Let stand in ice-box. (A). (B) allowed to evap., residue taken up in a little H_2O , filtered, washed w. $\frac{50}{2} alc.$ Filtrate C. B diss. w. sl. xs. dil. NH_4OH , filtered from traces insol. material, acidif. w. 0.2 cc. conc. HNO_3 . Free acid began to sep. almost immed. C b. to ~~remove~~ remove alc., cooled, ext. vol. of ca. 4 cc., made alk. w. 40% KOH . Heavy crystn. salt immed. sep. Centfgd. Ppt. C_1 . Supernatant C_2 . C_1 washed 1" w. ca. 1 cc. H_2O , then b. w. 2-3 cc. dil. H_2OAc . Most remained insol. C_2 acidified strongly w. H_2OAc to get KH saccharate if present. All let stand in ice-box.
- MAR 6 - A filtered on sm. Buchner, washed several times w. sm. amts. H_2O , then w. sm. amts. acetone. Yield: 0.21 g., m. & decomp. 216° w. prelim. soft. & blackening. MAR 7 - B filtered off, washed w. H_2O , acetone, dried. Yield: 0.070 g., m. & decomp. 215° w. prelim. soft. & blackening. C_1 filtered off w. aid of 50% alc., washed w. 50% alc., alc., acetone, dried. Yield: .191 g.
- C_2 filtered from sm. amt. crystals, probably same as C_1 , seeded w. KH saccharate.
- Mar. 9 - Remainder of A & all of B recrystd. from H_2O . Combined. Yield: 0.14 g., ^{recomp.} m. 216° , w. prelim. dark. & soft.
- NOV 12 1926 - C_2 decanted from small deposit of heavy crystals. These recrystd. from l.c. H_2O (C_2). ~~Def~~ Decanted C_2 (C_3) still. & b. w. lower black. Tested for opt. act. & found to contain strongly dextro rotatory subst. Rotation $+5.7^\circ$, 7 cc. $l=1$. Although strongly acid w. H_2OAc , gives ppt. w. basic Pb acetate. This added until no more permanent ppt.

SSS 39

NOV 22 - C₂ = ca. 12 mg. Slight quinoline-like odor on burning.
 Intumesces. MICROKJELDAHL on 9 cc. No N

C₃ - Basic Pb ppt. (before neutral.) centrifgd. off. Ppt. C₃ Supnt.
 (C₄) treated w. ex. basic Pb ac. & set in ice-box, centrifgd., ppt. suspd. in
 H₂O, decompd. w. H₂S. C₃ - also.

C₃ - after filtern. & removal of H₂S gives charac. ppt. of Ca C₂O₄ w.
 CaCl₂ + aq. NH₃.

C₄ - filtered & H₂S removed by air. at vol. ca 14 cc. in 2 dm.
 tube. Bl. - 03.02 .02 Rdp; +0.11 .09 .08 .09
 $\alpha = +0.11^\circ$

Concd. dry in vac. & evapd. several times w. H₂O. Yield: 0.025g.
 Added to con. portion of (62)

S.S.S. 39

Wt. bot. + subst. 8.1310
 " " 7.9241
 .2069

C + H by Gartmann,
 on oxidation product

COOH C₆ = 72.
 | (CHOH)₄ H₁₀ = 10.08
 | COOH O₈ = 128
 210.08 C = 34.27%
 H = 4.80% (theory)

MAR 10 - Recrystd. ^{A & B} Wt. bot. + subst. 8.0351
 " " 12 7.9241
 .111

0.1008 g. anhyd. subst. for combustion

CaCl₂ 63.8079 NaCl 55.4795
 63.7637 55.3574
 .0442 .1221
 .0032
 .1253

Guard 45.0685
 45.0653
 .0032

C = 33.9%
 H 4.9%

Fraction C₁
 Wt. bot. (8) + subst. 7.4265 7.4260 7.4258 7.4259
 " " " 7.2345 7.2345
 .1914

7.4259
 7.3259
 .1000

Dis. in 6 cc. H₂O by lit., cooling rapidly.

OPTICAL ROTATION

Bl. -0.07 .05 .06 .06 Rds. -0.03 .05 .05

Rinsed into weighed Pt dish w. H₂O, evapd. dry, ignited w. H₂SO₄.

Wt. dish + K₂SO₄ 23.4551
 " " 23.4001
 .0550

(8) Recrystd. from dil. HOAc.
 Wt. bot. + subst. 7.3020 7.3005
 " " 7.2347 7.3004
 7.2350
 .0654

Wt. Pt dish + K₂SO₄ 23.4325 23.4320
 " " 23.3995
 .0330
 .0325
 .0146

-COOK 36
 | H-C-OH 3
 | COOH 89
 39.1
 158.1
 K₂ 24.7%

Wt. K
 % K 22.3%

5'S'41

Type I

12/31/24	1/196/6	XII-24	12	1/2/25	1/3 1/3
1/2/25	1/196/6	XII-24	14	1/3	1/5
1/5/25	1/196/6	XII-24	14	1/6	1/7
1/6	1/196/6	XII-24	14	1/7	1/8
1/27	1/202/3	I-20	14	1/28	1/29
1/28	1/202/3	I-20	14	1/29	1/30
1/29	1/202/3	I-20	14	1/30	1/31
1/30	1/202/4	I-21	14	1/31	2/2
2/2	1/202/3,4,5	I-20,21,23	1,11,2	2/3	2/4
2/3	1/202/5	I-23	138/14	2/4	2/5
2/4	1/202,204/4,5,3	I-21,23,28	1,3,10	2/5	2/6
2/5	1/204)3	I-28	14	2/6	2/7
2/6	1/204/3	I-28	14	2/7	2/8
2/10	1/204/3,4	I-28, II-2,3	1,10,3	2/11	2/12
2/11	1/204/4	II-3	14	2/13	2/14
2/13	1/204/4,5	II-3,4	14	2/14	2/16

15
14
60
15
210
12
222

FLASKS 222
LITERS 333

FEB 16 1925 — Cryst. in concentrate

diss. by warming, cooled again w. ice,

& at vol. of 5 l., treated w. 1:1 HCl until blue to Congo. After standing 1 hr. in ice-box, greenish ppt. centrifgd. off in cold, washed 1" w. N/100 HCl. at vol. of 5500 cc., pptd. w. 8 l. alc. FEB 17 - Part of ppt. had not settled, so let stand 1 day longer while main ppt. colled. up by centfy. & washing w. 1/2 HOAc. FEB 18 - 2" ppt. treated same way, both combined, taken up in H2O, diss. by centfy. w. NaOH, & at vol. of 950 cc. pptd. in cold w. 400 cc. alc. FEB 19 - Supernatant - , so mixt. centrifgd. in cold.

2

3

HOAc

FEB 20 - Ppt. taken up in 1/2 HOAc, let stand ca. 2 hrs., centrifgd. Ppt.

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dis. w. just enough NaOH, 10 g. NaOAc added, & at vol. of 500 cc.,
 4 pptd. w. 300 cc. alc. Let stand in ice-box. FEB 21st Supernatant -,
 H₂OAc mixt. centrifgd., ppt. taken up in N/2 HOAc. Let stand over night,
 centrifgd., ppt. taken up in H₂O, neut. w. NaOH. Made alk. w.
 10 cc. 10% NaOH, 10 g. NaOAc added, & at vol. of 330 cc., pptd. w.
 5 400 cc. alc. Still ++, so 300 cc. more alc. used. Still ++. 200 cc.
 NaOH Et₂O added, completing the pptn. Centrifgd. Yellow supernatant.
 Ppt. taken up in H₂O, centrifgd., & at vol. of 260 cc., treated w. 10 g. NaOAc,
 6 10 cc. 10% Na₂CO₃, & pptd. w. 500 cc. alc. Supernatant -, overnight in
 Na₂CO₃ HOAc cold. Ppt. centrifgd. off, taken up in 200 cc. H₂O, 12 cc. HOAc added. MAR 2 -
 Ppt. centrifgd. off, taken up in H₂O, dis. by making slightly alk., centrifgd.
 7 from considerable insol. material. MAR 3 - 20 g. NaOAc added.
 NaOH MAR 4 - 5 cc. 10% NaOH added, & at vol. of 230 cc., pptd. w. 500 cc. alc.
 8 MAR 5 - Supernatant -, sucked off, ppt. centrifgd., taken up in H₂O.
 Ba(OH)₂ Centrifgd. Mar 9 - ^{Cold} soln. pptd. w. xs. Ba(OH)₂ soln. (hot, concd.), stirred,
 let stand in ice-box. 10 - Most of color in supernatant. Centrifgd.
 Ppt. stirred up w. H₂O. MAR 11 - St. xs. strong H₂SO₄ added, all lumps
 disintegrated, ppt. centrifgd. off. Supernatant (A) contained
 very little S. Ppt. again worked w. dil. HCl. MAR 12 - This time
 S came out & nec. to dil. very viscous soln. to ca. 125 cc. before could
 be centrifgd. Ppt. washed 2 x w. ca 40 cc. H₂O acidif. w. HCl. Combined
 extd. centrifgd. from trace BaSO₄, soln. made d. alk. w. NaOH, pptd.
 9 by adding 13 cc. HOAc. MAR 13 - Centrifgd. repeatedly, supernatant -,
 H₂OAc discarded. MAR 14 - Ppt. taken up in H₂O, dis. w. d. xs. Aq. NH₃, (N/2) CO₂Y
 added to remove any Ca present. MAR 15 - Trace of ppt. centrifgd. off,
 supernatant acidif. to Congo w. 1:1 HCl, & at vol. of ca. 125 cc., pptd. w.
 10, 11 250 cc. redistd. alc. Centrifgd. Ppt. taken up in H₂O + ~~little~~ HCl, reprecip. w. ^{redistd.} alc.
 HCl Centrifgd. Ppt. taken up in 50% alc. Part dissolved. MAR 17th All lumps
 smoothed out, made up to 250 cc. w. ~~redistd.~~ redistd. alc. Centrifgd. Ppt. taken up
 in a little H₂O, let stand 0.5 hr., completely pptd. by dilg. to 250 cc. w.
 redistd. alk. Centrifgd. after 1 hr. Supernatant contained much Cl⁻. Process
 repeated twice, then ppt. taken up in H₂O alone several times, washings
 being only feebly ±, finally washings & ppt. Cl⁻ free. Taken up in ~~alc.~~ ^{acetone} ~~the~~
 redistd. acetone several times, filtered off, washed w. acetone, dried.
 Yield: 2.5 g. H₂O soln. Active at 1:5,000,000

MAR 31 - When titrated to alkaly. w. phenolphthalein, formaldehyde im-
 med. discharges color. 10 mg. req. 0.52 cc. N/100 NaOH to bring it back again.
 Soln. still sp., but does not give ppt. on taking part former insol. ph. w.
 acid.

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MAR 27 - Wt. bot (6) + subst. 7.3769 7.3768
 7.2184
 .1584

Diss. w. H₂O + dil. H₂SO₄,
 next, w. NaOH until just
 back in soln. again, made
 up to 31.68 cc. w. saline.

ASH

Wt. bot + subst	7.2187	7.1508	Wt. Pt dish + ash	23.4000
" "	7.1508	7.1038	" " "	23.4000
	.0679	.0470		

Unweighable trace of ash.

No Ash

OPTICAL ROTATION

MAR 31 Blanks: -0.07 .08 .08 .08 Rdg: +1.40 1.45 1.44 1.43 l = 1
 $[\alpha]_D = + \frac{(1.44 + 0.075)}{1} \times 2 = +303^\circ$

**$[\alpha]_D$
= +303**

MICROKJELDAHLS

2 cc. samples

N/50 HCl (x 1.003)	3.00	3.00	6.00
N/100 NaOH (x 0.998)	8.11	10.41	2.21
1.00%	6.01	8.11	3.81
	2.10	2.30	.55
			3.26

3.26 x 0.14 = .46

**Total N
4.6%**

AMINO ~~FORNICE~~ TITRATION NITROGEN

APR 3

T	P	Rdg.	Rdg. -
24°	759.5	0.15	Bl.
		.16 ¹⁵⁵	Bl.
22°	"	.640	1
20.5°	760	.600	2

$\frac{.485 \times .5631 \times 100}{2} = 13.65 \text{ mg.}$
 $\frac{.445 \times .5678 \times 100}{2} = 12.63 \text{ mg.}$
 236.28
 13.14 mg.
 per 500 mg.

Jun 20 - Repetition of Total N by distn.

**Amino N
(NO₂ Method)
2.6%**

Wt. bot + subst	7.3181	7.2980
	7.2980	7.2758
	.0201	.0222

cc N/50 HCl (x 0.995)	15.00	15.00
" " NaOH (x 1.015)	9.60	9.15
	14.92	14.92
	9.74	9.29
	5.18	5.63
	.11	.11
	5.07	5.52
	1.014	1.104

5.05 4.97

**Total N
5.0%**

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APR 2-

HYDROLYSIS w. HNO₂

4 cc. soln. + 6 cc. ~~10%~~ 30% NaNO₂ soln. + 3.5 cc. HOAc occasion-
ally shaken for 30 min. 1.2 cc. 18 N H₂SO₄ added & HNO₂ aspirated
out for 30 min. Set stand in ice-box.

APR 3-

Made up to 25 cc.

OPTICAL ROTATION

$l = 2$ $d = 0.03$ $\alpha = 0.01702$ Rdgs. +.26 26 .28 .26

$$[\alpha]_D = + \frac{.28 \times 100}{2 \times .08} = +175^\circ$$

Bl. 19.20 (gub) 19.50
19.00
14.35

Redg. Sugars

3 cc. samples
Rdgs. 14.35
Redg. sugars 32.5%

32.5%
Redg. Sugars

$4.65 \times 1.02 \times .318 = 1.54 \text{ mg. Cu} = 0.78 \text{ mg. glucose}$

~~Sugar sample~~
.8 x 3 mg. in sample

Bl. 19.00
16.13

Rdgs. 16.00 16.25

$2.87 \times 1.02 \times .318 = .93 \text{ mg. Cu} = 0.55 \text{ mg. glucose}$

23.0%

APR 6-

Started again w. 3 cc. soln. + 4 cc. 30% NaNO₂ soln.
& 2 cc. HOAc. 0.8 cc. 18 N H₂SO₄ added & HNO₂ aspir-
ated 1 1/4 hrs. Made up to 25 cc. had 10 cc. reagent by mistake.

Bl. 38.00
32.75

38.00
32.65

5 cc. sample

$5.35 \times 1.02 \times .318 = 1.74 \text{ mg. Cu} = 0.87 \text{ mg. Cu}$

Repeated w. 5 cc. reagent.

Bl. 19.00 19.00
14.30 14.30

$4.70 \times 1.02 \times .318 = 1.52 \text{ mg. Cu} = 0.78 \text{ mg. glucose}$
in 3 mg. sample

Redg. Sugars
26.0%

HYDROLYSIS w. HCl + SnCl₂.

APR 7 - 2 cc. soln. + 2 cc. H₂O + 4 cc. 2 N HCl + few crys. SnCl₂. 6.
7 hrs. under rf. Color remained faint pink. APR 8 - Dil. to 25
cc., H₂S passed in, centfgd., aspirated, 5 cc. taken for sugar anal.

Bl. 19.00
17.00

19.00
17.00

$2.50 \times 1.02 \times .318 = .65 \text{ mg. Cu} = 0.40 \text{ mg. glucose}$ in 2 mg. sample

Redg. Sugars
20.0%

Hydrolysis of S1 with HCl and Bromine -

100 mg of S1 was heated for 8 hrs with 15cc 2N HCl and an excess of Bromine. There was a vigorous evolution of CO_2 in the beginning as shown by the ppt of BaCO_3 in a trap on the condenser.